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Chapter Number: 2

Question type: Multiple choice

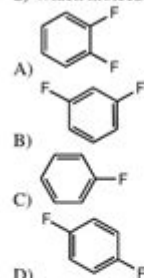
1) Which of the following compounds contains polar covalent bonds?

- A) CS_2
- B) LiF
- C) F_2
- D) CH_3F
- E) None of these choices.

Answer: D

Topic: Polar Covalent Bonds
Section Reference 1: 2.2
Difficulty: Easy

2) Which molecule does not have a dipole moment?



- E) None of these choices.

Answer: D

Topic: Molecular geometry, polarity
Section Reference 1: 2.3

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Answer: B Topic: Atomic Orbitals, Periodic Trends, Electronegativity Section: 1.3 Difficulty Level: Easy. 4) Select the least electronegative element from the list below.

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C) CCl4 D) BF3 Cl. Cl. E) Cl. Cl. Answer: A Topic: Molecular geometry, polarity Section Reference 1: 1.5, 1.6 and 2.3 Difficulty: Medium. 14) Of the following common organic solvents which one is ...

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