

KEY

16. What two common bases are used in the production of soaps? Sodium hydroxide (NaOH) and potassium hydroxide (KOH)
17. Give the name of an alkaline gas which is highly soluble in water. Ammonia (NH₃)
18. Nitric acid from the supplier is 16 M or 69 % HNO₃.
19. Which acid is a *non-electrolyte* in its concentrated form? acetic (CH₃COOH)
20. Drain cleaner is made up of mainly Sodium hydroxide (NaOH) (some has H₂SO₄)
21. Name two compounds that absorb CO₂ from the air. Sodium hydroxide (NaOH)
and potassium hydroxide (KOH)
22. Which acid is used in the manufacture of textiles? acetic acid (CH₃COOH)
23. Hydrochloric (HCl) acid is used to remove "scale" from boilers.
24. Name an acid which produces a lot of heat when mixed with water. Sulphuric (H₂SO₄)
25. Name a base which produces a lot of heat when added to water. Sodium hydroxide
26. Potassium hydroxide (KOH) is used as an electrolyte in alkaline batteries.
27. Write balanced formula equations for the reactions which happen when the following are mixed:
- a) sodium hydroxide and phosphoric acid
 $3\text{NaOH} + \text{H}_3\text{PO}_4 \rightarrow 3\text{H}_2\text{O} + \text{Na}_3\text{PO}_4$
- b) sulphuric acid and aluminum hydroxide
 $3\text{H}_2\text{SO}_4 + 2\text{Al}(\text{OH})_3 \rightarrow 6\text{H}_2\text{O} + \text{Al}_2(\text{SO}_4)_3$
- c) Fe(OH)₃ + HNO₃
 $\text{Fe}(\text{OH})_3 + 3\text{HNO}_3 \rightarrow 3\text{H}_2\text{O} + \text{Fe}(\text{NO}_3)_3$

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Note Taking Answers Acids Bases And Salts

ACIDS: ♦ A _____ taste is a characteristic property of all acids in aqueous solution. ♦ Acids react with some metals to produce _____ gas. ♦ Because aqueous acid solutions conduct electricity, they are identified as _____. ♦ Acids react with bases to produce a _____ and water. ♦ Acids turn _____ different colors. BASES:

11-04,05 Acids, Bases, Salts wkst

Some of the worksheets below are Acids and Bases Worksheet Middle School : Acid vs. Base Characteristics, Bronsted Acids and Bases, Identify Conjugate Acid/Base, Acids, Bases, and Conjugates : Recognizing Strong versus Weak Acids, Relationships between pH and pOH, ...

Acids and Bases Worksheet Middle School - DSoftSchools

Acids Bases And Salts Note Taking Answers *FREE* acids bases and salts note taking answers Acids, Bases, and Salts T3 G. Solutions of acids and bases 1. Acid describes compounds that can be ionized in water to form hydronium ions. 2. Base describes compounds that can form hydroxide ions in solution. 3. Solutions of acids and solutions of bases are electrolytic conductors to some ...

Acids Bases And Salts Note Taking Answers

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Note Taking Worksheet Acids Bases Salts Answers

NOTE PACKET Unit 9: Acids, Bases, & Salts *KEY* KEY * 2 Unit Vocabulary: Amphoteric Arrhenius acid Arrhenius base Bronsted-Lowry acid Bronsted-Lowry base Electrolyte hydronium ion hydroxide ion indicator (acid/base) neutralization pH scale titration Unit Objectives: Compare and contrast properties of acids, bases, and salts Compare the Arrhenius and Bronsted-Lowry theories of acids and bases ...

Mr. Dolgos Regents Chemistry NOTE PACKET Unit 9: Acids ...

Instructions Before viewing an episode, download and print the note-taking guides, worksheets, and lab data sheets for that episode, keeping the printed sheets in order by page number. During the lesson, watch and listen for instructions to take notes, pause the video, complete an assignment, and record lab data. See your classroom teacher for specific instructions.

Chemistry 1101: Introduction to Acids, Bases, and Salts ...

Acids and Bases - Neutralization Goals : To gain an understanding of : 1. The properties of acids and bases 2. pH and pOH calculations 3. Definitions of acids and bases 4. Neutralization reactions NOTES: Properties of acids : • Form hydrogen ions (H⁺ or

protons) in solution • Form hydronium (a water molecule bonded to a hydrogen ion shown as H₃O⁺ or H⁺. H₂O) ions in aqueous solution ...

CHEMISTRY NOTES - CHAPTERS 20 AND 21 Acids and Bases ...

Lecture Notes for Chapter 16: Acids and Bases . I. Acids and Bases. a. There are several ways to define acids and bases. Perhaps the easiest way to start is to list some of the properties of acids and bases. b. The table below summarizes some properties that will be helpful as we learn more about acids and bases. In general, though, it is ...

Lecture Notes for Chapter 16: Acids and Bases

Acids, Bases, and Salts ACID-BASE THEORIES Acids and bases are all around us and part of our everyday life (ex. bodily functions, vinegar, carbonated drinks, citrus fruits, car batteries, antacids, cleaning products, etc.). Acids and bases have the following properties: Acids Bases Tart or sour taste Bitter taste Electrolytes when in aqueous solutions (some are strong electrolytes, some are ...

Chapter 19 Acids, Bases, and Salts - Science with Mrs ...

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Bronsted-Lowry Conjugate Acids-Bases Pairs Acid Base H₂O NH₄⁺ HCO₃⁻-H₃O⁺ N H OH-NH₃ CO₃²⁻-H₂O N₂H₃⁺-H⁺ CHEM 0012 Lecture Notes 16 2 4 H₂O NH₂-HCO₃⁻-PO₄³⁻-H₃O⁺ NH₃ H₂CO₃ HPO₄²⁻-H⁺ 26/02/2012 9 Acids, Bases and Salts Hebden - Unit 4 (page 109-182) We will cover the following topics: 1. Amphiprotic Substance 2. Amphoteric Compounds CHEM 0012 Lecture Notes 17 Acids, Bases and Salts Hebden ...

Acids, Bases and Salts - Welcome to nobel.scas.bcit.ca

About This Quiz & Worksheet. The multiple-choice questions on this quiz and worksheet make it easy to find out how much you understand about household acids, bases and salts.

Quiz & Worksheet - Household Acids, Bases & Salts | Study.com

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Acids Bases And Salts Note Taking Worksheet Chapter 23 ...

Addition of Acids or Bases to Water. The process of dissolving an acid, specially nitric acid or sulfuric acid or a base in water is a highly exothermic one. As a rule: Always add acid to water and never the other way! The acid must be added slowly to water with constant stirring. If one mixes the other way by adding water to a concentrated acid, the heat generated causes the mixture to splash ...

acids bases and salts for class 10 cbse notes

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Hydroxide + chloride sodium ammonia + Bases Definition: Bases are metal oxides (also known as basic oxides) that react with acids to produce salts and water. Note: Most bases are insoluble except bases with Group I metals and ammonium ion (NH₄⁺); Soluble Bases I Insoluble Bases I Bases containing group I metals I Bases containing all other metals generally Bases containing ammonium ion (NH₄⁺) I ...

Chemistry Notes and Worksheet: Acids, Bases and Salt Essay ...

a chemical reaction between an acid and a base taking place in a water solution. Salt compound formed when the negative ions from an acid combined with the positive ions from a base; salts also form when acids react with metals. Salts are essential for many animals; Example- Salt blocks cow lick. Manufacturing of paint, rubber, glass, soap, detergents, and dry cell batteries are all ...

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