

Chemistry _____ Name _____
 _____ Per _____

ionic Salts Simulation Lab

Learning Goals

- Understand how an ionic salt dissolves describing a saturated solution microscopically and macroscopically.
- Calculate solubility in grams/100ml
- Determine why some salts are more soluble than others
- Design and carry out a virtual lab to test a hypothesis

Directions: Go to http://phet.colorado.edu/simulations/new.php?sim=Salts_and_Solubility and use the Salts and Solubility simulation. Take a few minutes to get use to how the simulation works. Then answer the questions below.

Questions:

- Describe what you observe when you shake AgCl into water. Test numbers of particles with different volumes.
 - When the salt dissolves, which element becomes the anion (the negatively charged atom) and which one becomes the cation (the positively charged atom)?
 - What is the relationship between amount of particles and amount of water? (direct or indirect)
 - At 5.00×10^{21} what is the point of saturation (point where you can't dissolve any more salt)?
- Fill in the chart below to see if the ratio of particles to water is constant at the saturation point.

Particles	Water Volume	Ratio (particles/volume)

 - Calculate the constant below (show your work)
- Test a variety of slightly soluble salts, noting the relationship between numbers of each ion pair. Fill in the chart below to help you see a relationship.
 - How many particles of each element need to be present before the salt stops dissolving (point of saturation)?

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Question: DATA SHEET B Name Lab And Space Solubility In: Compounds H2O Et2O 5% NaOH 5% NaHCO3 5% HCl Conc. H2SO4 Comments 1. CH3CO2H; Acetic Acid S S (NOT TESTED SINCE S - Low C Number, Polar. Should Dissolve In H2O. E.g.(expected) PH

Solved: DATA SHEET B Name Lab And Space Solubility In: Com ...

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Chemistry 1002: Solubility | Georgia Public Broadcasting

Question: Lab Report Sheet: Name: Lab Section (Day And Time) Chemistry 1020 Lab 10 Solubility Of KNO, Pre-lab Exercises: What Type Of Solution Exists If There Is Excess, Undissolved Solid Present? A. Saturated B. Supersaturated C. Unsaturated D. Non-electrolyte Extrapolation Of Data Is To A. Export It To A Spread Sheet C. Make Labs Longer B. Predict A Data Point ...

Solved: Lab Report Sheet: Name: Lab Section (Day And Time ...

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EXPERIMENT 1: SOLUBILITY TESTING Data Sheet Table 1. Hydrocarbon Solubility Test Results Hydrocarbon Solubility in Water (Yes/No) Solubility in Heptane (Yes/No) Heptane NA 1-Octadecene Additional Observations: Experiment 1 Post-Lab Questions 1.

Solved: EXPERIMENT 1: SOLUBILITY TESTING Data Sheet Table ...

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Lab: Solubility Assignment: Reflect on the Lab Flashcards ...

Solubility Lab Report Results 2 Spoonfuls The sugar fell to the bottom of the beaker. After 30 seconds, the majority was dissolved, but there was still some sugar left undissolved. 1 Spoonful Once the sugar was poured into the water, it fell to the bottom of the beaker. After 10

Solubility Lab Report by Sarah Arndt - Prezi

Question: Lab 4 Data Sheet: Identity Of A Liquid Name Die Instructor Time & Day Of Lecture PART A. Accene Data Mass Of Graduated Cylinder Trial 1 Volume Of First Sample... Mass Of First Sample And Graduated Cylinder Mass Of First Sample Density Of First Sample: Show Calculation 33.9409 Somu Ocelone) U09419 Trial 2 Total Volume Of Second Sample Som Mass Of Second ...

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